Reaction of Monosaccharide Derivatives with Stabilized Sulfur Ylides. A Highly Stereoselective Synthesis for C-Glycofuranosides

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Abstract: Stereoselectivity in the formation of glycidic amides by reaction of various aldehydo- and ketosugars [(1), (4), (6), (8), (10) and (12)] with N,N-dimethyl-2-(dimethylsulfuranylidene)acetamide (2) was studied. The reaction with derivatives of reducing cyclohemiacetalic monosaccharides (14) and (21), gives to α -Cglycofuranosyl-2(S)-hydroxyacetamides with a higher yield and stereoselectivity.

INTRODUCTION

We recently developed a highly stereoselective synthesis for glycidic amides (3) by reaction of 2,3-Oisopropylidene-D-glyceraldehyde (1) with N,N-dimethyl-2-(dimethylsulfuranylidene)acetamide (2), which is virtually quantitative¹



This provides a new, ready procedure for incorporating two new chiral carbons into an acyclic carbon chain. In addition, the resulting epoxide function is one of the most useful in organic synthesis, on account of the high synthetic versatility of epoxides and the regioselectivity that may arise from the presence of the amide group in subsequent reactions.

The results led us to study the versatility and potential uses of the process. Thus, in a first step we extended the reaction to other *aldehydo*-sugar derivatives and one ketose (Scheme 1). Next, we studied the reaction with two derivatives of reducing cyclohemiacetalic sugars (Scheme 2 and 3).

RESULTS AND DISCUSSION

The reaction of 2,3-O-isopropylidene-D-glyceraldehyde with ylide 2 in chloroform yields *trans*-2,3epoxyamides 3a and 3b (86:14 at r. t., 96:4 at -5 to 0°C) with high diastereoselectivity. In order to examine the solvent effect on the stereoselectivity, we implemented the reaction in various solvents (CH₃CN, EtOH, THF, benzene) and also in their absence (by simply mixing the reactants); the ratio of the two stereoisomers,

Scheme 1

Reaction of sugar derivatives 1, 4, 6, 8, 10 and 12 with ylide 2 at room temperature

Sugar derivative	2,3-epoxyamide	stereoisomer ratio*	solvent	yield ^b
	_	86:14	CHCl ₃	92%
to	10	96:4*	CHCl ₃	c.a.100%°
	CONTRACT CONTRACT	86:14	CH, CN	66
01-0	8 -	87:13	EtOH	**
1	3a,b	83:17	THF	**
	_	85:15	Benzene	46
Lo ch-o	To contine	87:13	without	95%
4 Me Lo CH=0	5a,b	86:14	without	95%
6 Linno	7a,b 7a,b 7a,b 7a,b 7a,b	3:2	CHCl,	94%
**** **** ****************************	Sa,b 9a,b 9a,b 9a,b	3:2	CHCl,	86%
10	Acc OAc 11a,b	<i>c.a.</i> 3:1	CHCI,	11%
د. المعاد الم 12		<i>cis-</i> isomer <i>ratio</i> 4.5:20	CHCl ₃	95%
	^{13c,d} 13c,d	trans-isomer ratio 22:53.5	-	

*Determined by NMR and GC/MS; b Yield of pure isolated products; Vield determined by GC/MS; at -5 to 0°C

as determined by ¹H MNR and GC/MS was nearly always the same. We obtained identical results for the reaction between 2,3-O-isopropylidene-L-glyceraldehyde (4) and ylide 2, which yielded the corresponding *trans*-2,3-epoxyamides **5a,b** (L-*arabino*:L-*xylono*, 86:14).

The reaction of 2,4-O-ethylidene-D-threose (6) with ylide 2 in chloroform at room temperature, was completed in a few minutes and a mixture of the two *trans*-2,3-epoxyamides 7a,b in a 3:2 ratio was obtained with a virtually quantitative yield. The isomers were isolated by flash chromatography and recrystallized.

The 2,3:4,5-di-O-isopropylidene-D-arabinose (8) was also reacted with ylide 2 under similar conditions to give a 3:2 mixture of the *trans*-2,3-epoxyamides, 9a,b with an overall yield of 86% (the more polar isomer was predominant). The individual isomers were isolated by flash chromatography and the minor isomer was obtained in crystalline form.

Only the *trans*-isomers were formed in both cases ($J_{23} = 2.1-2.2$ Hz), but not the desired stereoselectivity. The ratio of stereoisomeric products was nearly always the same, but their absolute configuration could not be established.

As shown in previous work¹, the high stereoselectivity observed in the reaction of 2,3-O-isopropylidene-Dglyceraldehyde (1) can be ascribed to the ylide approaching the less hindered face (the *si* face) of the carbonyl group according to the preferred Felkin-Anh model. As can be readly seen with the aid of molecular models, the presence of a second dioxolane group in **8**, introduces additional hindrance over the *si* carbonyl face, leading to a lesser preference for this approach, which results in a change in the above mentioned proportion for 1 to 3:2 for 7 and 9.

The poor stereoselectivity achieved with compound 6, may arise from the occurrence of an intra or intermolecular hydrogen bond between the 3-OH function and the carbonyl group that alters the reaction course.

The reaction of 2,3,4,5-tetra-O-acetyl-D-xylose (10) with ylide 2 was seemingly quite fast, but the clear initial solution turned dark and fairly small amounts (11%) of the desired epoxides were obtained. ¹H NMR showed both epoxides 11a,b to have a *trans* configuration about the oxirane ring ($J_{23} = 2$ Hz).

The reaction of 1-deoxy-3,4-O-isopropylidene-D-glycero-2-tetrulose (12) with the ylide 2 completed in c.a 1 h. GC/MS analysis of crude reaction products showed a clean mixture of the four possible diastereoisomeric epoxides 13a-d, with an essentially quantitative yield, and in a ratio, in order of increasing retention time, of 4.5: 20:22:53.5, respectively. No stereochemical assignment of these compounds was attempted. Partial chromatographic separation of isomers 13a and 13b was possible. Isomers 13c and 13d were isolated in analytically pure form, but not on the preparative scale. By comparison of NMR spectra, we believe isomers 13a and 13b are both cis, while isomers 13c and 13d, are *trans*. The main differences between the 13a,b and 13c,d pair, lie in the chemical shift of H-2, C-2 and C-3:

	13a	13b	13c	13d
H-2	3.41	3.41	3.68	3.57
C-2	60.1	61.3	56.9	58.3
C-3	60.2	61.5	61.2	61.3

TABLE 1.- Major chemical shifts of the isomers of compound 13

The mass spectra of isomers 13a and 13b also showed an intense m/z 157 fragment (M- CONMe₂). Interestingly, this fragment was not observed in the tentatively assigned E isomers 13c and 13d.

The high reactivity observed in free carbonyl groups prompted us to extend the reaction to other, less reactive sugar derivatives, such as some of reducing cyclohemiacetalic sugars [2,3-O-isopropylidene-D-ribofuranose (14)(Scheme 2) and 2,3:5,6-di-O-isopropylidene-D-manofuranose (21) (Scheme 3)], in the hope that the minor aldehyde form present in the equilibrium mixture would suffice for the reaction works. Subsequent cyclization by intramolecular opening of the oxirane ring might provide a new, regio- and stereoselective approach to C-glycofuranosides.



Ribose derivative 14 was reacted with the ylide 2 in chloroform. After 2h at 60°, TLC analysis revealed the absence of ylide and the presence of residual starting sugar and a sole new product in a *c.a.* 1:1 ratio. Examination of the ¹H and ¹³C NMR spectra of the crude mixture, revealed the absence of epoxides. Thus, we assumed that the initial reaction product had undergone intramolecular cyclization *via* oxirane ring opening to give a C-glycoside. The same product was obtained by reaction at room temperature, when excess ylide was added sequentially until TLC confirmed depletion of the starting sugar. Only one glycoside was isolated (85 % yield) which was characterized spectroscopically and by comparison with similar compounds²³, such as α -C-glycoside 17. The ring size was determined from the ¹³C-chemical shifts of C-6 and C-7⁴, and the "anomeric" configuration from the J_{5,6} ¹H-NMR coupling constants of the tetrahydrofurane ring protons and the ¹³C-chemical shifts of the isopropylidene groups in the C-glycosides derived from 2,3-O-isopropylidene-D-ribose, are correlated with their anomeric configuration (Table 2). The β-anomer (18) was not directly detected; however, acetylation of the crude reaction mixture gave a minor acetylated product (<10%), which was characterized as a β-anomer (20) by ¹H- and ¹³C NMR spectroscopy, in addition to the major derivative (19). On thorough examination of the NMR spectra of the crude recorded in several runs, however, it became apparent that β-C-glycoside 18 was produced, although in much lesser amounts; we could not isolate the pure product, though.

	α-anomer		β-anomer	a-anomer	β-anomer	
	17	19	20	lit. ²	lit. ²	lit. ³
J.,	1 Hz	0 Hz	3.5 Hz	0-1 Hz	4.5 Hz	3.5 Hz
0- <u>C</u> -Ő	112.1	112.8	114.1	112.7±0.6	114.5±0.6	113.6-114.0
C-Me,	26.5	26.4	27.4	26.3±0.2	27.5±0.2	27.4-27.7
2	24.5	24.9	25.5	24.9±0.2	25.5±0.2	25.6-26.0

TABLE 2. Relevant spectroscopic data for C-glycosides obtained from 2,3-O-isopropylidene-D-ribose

Also, the ¹³C NMR spectra showed C-2, C-3 and C-4 in the α -anomer (19) (with the aglycon and its C4hydroxyl in *cis*) to appear upfield from those in the β -anomer (20) (with a *trans* arrangement)². The configuration at C-2 was assigned on the basis of the assumed reaction mechanism, via the initial formation of the α -*trans*-epoxide and subsequent stereospecific intramolecular opening of the oxirane ring with simultaneous formation of the tetrahydrofurane ring, in accordance with the previous reaction sequence. The intermediate epoxides 15 and 16 were not detected in this reaction; rather one *trans*-epoxide was the main product in the analogous reaction of the mannose derivative 21.

Scheme 2



Thus, the reaction of D-mannose derivative 21 with ylide 2 in chloroform at room temperature, gave a mixture of one *trans*-epoxide and the two anomeric C-glycosides (3.1:1.0 α/β ratio), in essentially quantitative yield; the α/β ratio depends on the reaction time; as is showed later, 22 cyclise to 24, increasing the α/β ratio to 6.5:1.0; the products were characterized as 22, 24 and 25. The *trans* configuration of the oxirane ring and that of the anomeric centre, were assigned on the basis of the NMR data for these compounds, as were those of the acetylated derivatives, which were prepared not only to provide more data for comparison, but also to achieve a higher ¹H NMR resolution in the protons.





TABLE 3.- Relevant spectroscopic data for Epoxides and C-glycosides obtained from 2,3:5,6-di-Oisopropylidene-D-mannose

	trans-epoxides		a-anomers		β-anomers	
	22	26	24	27	25	28
J ₂₃	2.0 Hz	2.1 Hz				
J_14			<1 Hz	0 Hz	3.5 Hz	1.2 Hz
Ĉ,	51.5	49.3	70.7	72.3	64.6	66.2
Ċ,	57.6	55.6	85.1	83.2	83.7	80.5
C ₄			81.4	81.1	80.8	80.2

The α -anomer of C-glycosides obtained from 2,3:5,6-di-O-isopropylidene-D-mannose, appears with a $J_{3,4}$ 0 Hz; this value is consistent with a *trans* arrangement of H-3 and H-4². These compounds do not allow a goc correlation between the ¹³C chemical shifts of the isopropylidene groups and the anomeric configuration Nevertheless, the β -anomers showed chemical shifts for C-2, C-3 and C-4 that were upfield of those of the c anomers².

The involvement of epoxide 22 in the formation of the final C-glycosides was confirmed by the cyclizatic of 22 to the major α -C-glycoside 24, which confirms the mechanistic basis for the configurational assignments C-2 in these products.



The high stereoselectivity observed in these reactions may be the result of rapid trapping of the open form of the sugar by ylide 2. Thus, it appears that the incipient carbonyl group retains the conformation of the preceeding hemiacetalic form of the sugar, as shown in structures 14A and 21A. One other explanation can be provided by using Molecular Mechanics (InsightII program) to minimize the anionic form of the open initial sugar derivatives, which must result from the basic reaction medium. Thus, starting from structures like 14A and 21A, interactive minimization shows the repulsive interaction between the oxi-anion and the oxo-(carbonyl) group until the corresponding conformational minimum is reached. These minimum energy conformations are also obtained from many other different intial conformations, appearing as the absolute minimum conformations. As can be seen from the stereographs, the less hindered face is the α -face (*si*-face), in both cases.



14A

EXPERIMENTAL

All mp values are uncorrected. Optical rotations were measured at 20-22°C with a Perkin-Elmer mod. 241 polarimeter. Low resolution mass spectra in the electron impact (EIMS) mode were recorded on a HP-5988; GC/ MS analyser using HP-1 Crosslinked Methyl Silicone Gum strips of 12m x 0.2mm x 0.33mm; the temperature program was 80 °C : 0.80 min, 30°C/min, 250 °C : 10 min. ¹H and ¹³C NMR spectra were recorded on a Bruker WP 200 SY spectrometer; Proton chemical shifts are referred to the residual chloroform (7.24 ppm), and carbon chemical shifts to the solvent (¹³CDCl₃= 77 ppm). ¹H- and ¹³C NMR signals were assigned from 2D COSY, 2D-NOESY, HCCORR and DEPT experiments. TLC was carried out on 60 F 254 silica gel plates and column chromatography was done on silica gel 60 (70-230 mesh or 230-400 mesh).

N,N-Dimethyl-2,3-anhydro-4,5-O-isopropylidene-L-*arabino*- and L-*xylono*- pentonamides (5a and 5b).

N,N-Dimethyl-2-(dimethylsulfuranylidene)acetamide (2, 0.62 g, 4.23 mmol) was added to 2,3-Oisopropylidene-L-glyceraldehyde⁵ (4, 0.5 g, 3.85 mmol) at room temperature. The reaction was complete in a few minutes. ¹H NMR and GC/MS analysis of reaction crude showed a clean mixture of two isomers in essentially quantitative yield. (5a, $T_R = 4.73 \text{ min}$, 86%; 5b, $T_R = 4.83 \text{ min}$. 14%). Analytical and spectroscopic data for 5a and 5b were identical to those for the previously reported¹ enantiomers 3a (D-*arabino-*) and 3b (D-*xylono-*).

Isomer **5a** was isolated by column chromatography on silica gel, using 1:1 hexane-ethyl acetate as eluent. $[\alpha] = +2.8^{\circ}$ (c 0.93, CHCl₃); ¹³C NMR (CDCl₃) δ : 166.3 (CO), 110.1 (OCO), 75.0 (C₄), 66.9 (C₅), 57.6[•] (C₃), 52.0[•] (C₂), 36.3 and 35.6 (NMe₂), 26.4 and 25.0 (C<u>Me₂</u>).

* These assignations were erroneously exchanged for the reported¹ D-isomer.

Anal. calcd. for C₁₀H₁₇NO₄: C, 55.80; H, 7.96; N, 6.51. Found: C, 55.84; H, 8.24; N. 6.87.

N,N-Dimethyl-2,3-anhydro-4,6-O-(S)-ethylidene-D-ido- and D-galacto-hexonamides (7a and 7b).

N,N-Dimethyl-2-(dimethylsulfuranylidene)acetamide (2, 1.14 g, 7.7 mmol) was added to a solution of 2,4-O-ethylidene-D-threose⁶ (6, 1.02 g, 7 mmol) in $CHCl_3$ (20 ml) at room temperature. The reaction was monitored by TLC and was complete in a few minutes. The solvent was evaporated to an amorphous white solid, which was showns by ¹H NMR to consist of two *trans*-epoxides in a 3:2 ratio. The isomers were separated by flash chromatography on silica gel using 10:1 ethyl acetate-methanol as eluent, followed by crystallization from methanol.

Major isomer (less polar): 905 mg (56 % yield); m.p. 93-96°C; $[\alpha] = -19^{\circ}$ (c 0.2, MeOH). ¹H NMR (CDCl₃) δ : 4.72 (q, 1H, J= 5.1 Hz, <u>CH</u>CH₃), 4.05 (dd, 1H, J_{e5}=2.0 and J_{e6}= 12.0 Hz, H₆), 3.85 (dd, 1H, J_{e6}= 12.0 and J_{e5}= 1.4, H₆), 3.81 (dd, 1H, J₄₅= 1.2 and J₄₃= 3.5 Hz, H₄), 3.75 (d, 1H, J₂₃= 2.1 Hz, H₂), 3.65 (ddd, 1H, J₅₄= 1.2, J₅₆= 1.4 and J₅₆= 2.0 Hz, H₅), 3.34 (dd, 1H, J₃₂= 2.1 and J₃₄= 3.5 Hz, H₃), 2.00 (sb, 1H, -OH), 3.15 and 2.96 (2s, 6H, NMe₂), 1.33 (d, 3H, J= 5.1 Hz, <u>CH₃CH</u>). ¹³C NMR (CDCl₃) δ : 166.8 (CO), 99.5 (OCO), 77.3 and 64.1 (C₄ and C₅), 71.7 (C₆), 56.6 (C₃), 51.3 (C₂) 36.3 and 35.7 (NMe₂), 20.8 (CH₃). EIMS *m/z* (%): 230 (M⁺-1, 1%), 159 (6), 144 (60), 114 (45), 87 (29), 72 (100), 45 (57), 44 (55), 43 (48).

Anal. calcd. for C₁₀H₁₇NO₅: C, 51.94; H, 7.41; N, 6.06. Found: C, 51.68; H, 7.53; N, 6.01.

Minor isomer (more polar) : 610 mg (38 % yield); m.p. 213-215°C (MeOH); $[\alpha] = -35^{\circ}$ (c 0.2, MeOH). ¹H NMR (CDCl₃) & 4.75 (q, 1H, J = 5.1 Hz, CH₃<u>CH</u>), 4.03 (dd, 1H, J_{e5} = 2.0 and J_{e5} = 12.0 Hz, H_e), 3.81 (dd, 1H, J_{e5} = 12.0 and J_{e5} = 1.4 Hz, H_e), 3.75 (dd, 1H, J₄₃ = 4.2 and J₄₅ = 1.4 Hz, H₄), 3.73 (d, 1H, J₂₃ = 2.2 Hz, H₂), 3.58 (ddd, 1H, J₅₄ = 1.4, J₅₅ = 1.4 and J₅₅ = 2.0 Hz, H₅), 3.37 (dd, 1H, J₃₂ = 2.2 and J₃₄ = 4.2 Hz, H₃), 3.14 and 2.96 (2s, 6H, NMe₂), 1.68 (sb, 1H, OH), 1.36 (d, 3H, J = 5.1 Hz, <u>CH₃</u>CH). ¹³C NMR (CDCl₃) & 166.6 (CO), 99.9 (OCO), 74.9 and 64.6 (C₄ and C₅), 71.9 (C₆), 57.2 (C₃), 50.3 (C₂), 36.3 and 35.7 (NMe₂), 20.8 (CH₃). EIMS *m/z* (%): 230 (M⁺-1, 1.5%), 159 (5), 144 (48), 114 (42), 87 (37), 72 (100), 45 (49), 44 (54), 43 (50). Anal. calcd. for C₁₀H₁₇NO₅: C, 51.94; H, 7.41; N, 6.06. Found: C, 51.60; H, 7.52; N. 6.05.

N,N-Dimethyl-2,3-anhydro-4,5:6,7-di-O-isopropylidene-D-glycero-D-ido- and D-glycero-D-galactoheptonamides (9a and 9b).

N,N-Dimethyl-2-(dimethylsulfuranylidene)acetamide (2, 0.8 g, 5.4 mmol) was added to a solution of 2,3:4,5di-O-isopropylidene-D-arabinose⁷ (8, 1.15 g, 5 mmol) in CHCl₃ (20 ml) at room temperature. The reaction was monitored by TLC and was complete in a few minutes. The solvent was evaporated and the residue purified by chromatography on silica gel using 1:1 ethyl acetate-hexane as eluent. A white solid (1.35 g, 86 %) was isolated, which was shown by ¹H NMR and GC/MS analysis ($R_r = 6.28 \text{ min and } R_r = 6.19 \text{ min}$) to be a mixture of two *trans*epoxides in a 3:2 ratio.

Anal. calcd. for $C_{15}H_{25}NO_6$: C, 57.12; H, 7.99; N, 4.44. Found: C, 57.07; H, 8.19; N, 4.29. Flash chromatography provided enriched mixtures of each isomer. The minor isomer (less polar) was isolated as white crystals by crystallization with ethyl acetate-hexane. The major isomer (more polar) was obtained by preparative TLC as a colorless syrup which crystallized spontaneously.

The major isomer (more polar) had: $[\alpha] = -4.17^{\circ}$ (c 0.34, CHCl₃). ¹H NMR (CDCl₃) &: 4.13 (dd, 1H, J₄₅= 7.0 and J₄₃ = 3.1 Hz, H₄), 4.1-3.9 (m, 3H, H₅, H₆ and H₇), 3.82 (dd, 1H, J= 7.7 and 7.0 Hz, H₇), 3.74 (d, 1H, J₂₃= 2.2 Hz, H₂), 3.42 (dd, 1H, J₃₂ = 2.2 and J₃₄ = 3.1 Hz, H₃), 3.13 and 2.96 (2s, 6H, NMe₂), 1.39, 1.38, 1.37 and 1.32 (4s, 12H, 2 x CMe₂). ¹³C-NMR (CDCl₃) &: 166.7 (CO), 110.3 and 109.9 (2 x OCO), 78.6, 78.3 and 76.8 (C₄, C₅ and C₆), 67.6 (C₇), 57.5 (C₃), 51.1 (C₂), 36.3 and 35.7 (NMe₂), 29.7, 27.0, 26.7 and 25.2 (2 x C<u>Me₂</u>). EIMS, *m*/ *z* (%): 300 (M*-15, 92), 214 (14), 128 (29), 101 (31), 72 (100).

The minor isomer (less polar) had: m.p. 92-94 °C; $[\alpha] = + 16.0^{\circ}$ (c 0.2, CHCl₃). ¹H NMR (CDCl₃) δ : 4.15-4.00 and 3.95-3.85 (2m, 4H, H₅, H₆, H₇ and H₇), 4.03 (dd, 1H, J₄₅ = 7.2 and J₄₃ = 3.4 Hz, H₄), 3.70 (d, 1H, J₂₃ = 2.1 Hz, H₂), 3.36 (dd, 1H, J₃₄ = 3.4 and J₃₂ = 2.1 Hz, H₃), 3.14 and 2.96 (2s, 6H, NMe₂), 1.37, 1.34, 1.33 and 1.30 (4s, 12H, 2 x CMe₂). ¹³C NMR (CDCl₃) δ : 166.8 (CO), 110.4 and 109.9 (2 x OCO), 78.8, 78.4 and 77.2 (C₄, C₅ and C₆), 67.8 (C₇), 57.1 (C₃), 51.3 (C₂), 36.5 and 35.6 (NMe₂), 27.1, 26.7, 26.3 and 25.3 (2 x CMe₂). EIMS *m/z* (%): 300 (M⁺-15, 92), 214 (3), 128 (40), 101 (32), 72 (100).

N,N-Dimethyl-4,5,6,7-tetra-O-acetyl-2,3-anhydro-D-glycero-L-galo- and D-glycero-L-idoheptonamides (11a and 11b)

N,N-Dimethyl-2-(dimethylsulfuranylidene) acetamide (2, 0.5 g, 3.4 mmol) was added to a solution of 2,3,4,5-tetra-O-acetyl-D-xylose⁸ (10, 1 g, 3.14 mmol) in CHCl₃ (40 ml) at room temperature. The reaction seemed to proceed

quickly, but the clear initial solution turned dark. A dark gum was obtained on evaporating the solvent. Silica gel chromatography of an aliquot (600 mg) yielded only 70 mg (11 %) of the two *trans*-epoxides isomers (in a ratio of c.a. 3:1). Further preparative TLC (1:2 hexane-ethyl acetate) allowed the major isomer to be isolated to some extent.

Major isomer: yellowish oil; ¹H NMR (CDCl₃) δ : 5.43 (dd, 1H, J_{5,4}= 4.3 and J_{5,6}= 5.9 Hz, H₅), 5.30 (ddd, 1H, J₆₅= 5.9, J₆₇= 4.5 and J₆₇= 5.7 Hz, H₆), 4.85 (dd, 1H, J₄₅= 4.3 and J₄₃= 6.3 Hz, H₄), 4.30 (dd, 1H, J_{7,6}= 4.5 and J_{7,7}= 12.0 Hz, H₇), 3.95 (dd, 1H J_{7,6}= 5.7 and J_{7,7}= 12.0 Hz, H₇), 3.69 (d, 1H, J_{2,3}= 2.0 Hz, H₂), 3.23 (dd, 1H, J_{3,2}= 2.0 and J_{3,4}= 6.3 Hz, H₃), 3.09 and 2.93 (2s, 6H, NMe₂), 2.10 (s, 6H, 2 x CH₃CO), 2.03 and 2.02 (2s, 6H, 2 x CH₃CO). ¹³C NMR (CDCL₃) δ : 170.2, 169.7, 169.6, 165.7 and 165.6 (4 x <u>CO</u>CH₃ and <u>CO</u>NMe₂), 70.3, 69.4 and 69.2 (C₄, C₅ and C₆), 61.7 (C₇), 54.4 (C₃), 52.4 (C₂), 36.3 and 35.7 (NMe₂), 20.6 and 20.5 (4 x <u>CH₃CO</u>). EIMS *m/z* (%): 403 (M⁺, 0.1), 360 (M⁺-43, 0.5), 331 (0.7), 114 (44), 72 (51), 43 (100).

Minor isomer: spectral data for a mixture with the major isomer. ¹H NMR (CDCl₃) δ : 5.4-5.28 (m, 2H, H₅ and H₆), 5.1 (dd, 1H, J₄₅= 4.6 and J₄₃= 4.5 Hz, H₄), 4.28 (dd, 1H, J₇₆= 4.7 and J₇₇= 12.0 Hz, H₇), 4.02 (dd, 1H, J₇₆= 5.4 and J₇₇= 12.0 Hz, H₇), 3.49 (dd, 1H, J₂₃= 2.0 Hz, H₂), 3.37 (dd, 1H, J₃₂= 2.0 and J₃₄= 4.5 Hz, H₃), 3.09 and 2.95 (2s, 6H, NMc₃), 2.09, 2.07, 2.03 and 2.02 (4s, 12H, 4 x CH₃CO).

N,N-Dimethyl-2,3-annydro-4,5-O-isopropylidene-3-methyl-D-pentonamides (13a, 13b, 13c and 13d)

1-Deoxy-3,4-O-isopropylidene-D-glycero-2-tetrulose^o (12, 0.9 g, 6.3 mmol) was added to the N,Ndimethyl-2-(dimethylsulfuranylidene)acetamide (2, 1.2 g, 8.2 mmol) at room temperature. The reaction was monitored by ¹H NMR and was complete in *c.a* 1 h; at least three new compounds were among the products. GC/ MS analysis of crude reaction showed a clean mixture of four isomers in essentially quantitative yield. (13a, $T_R =$ 4.65 min, 4.5%; 13b, $T_P = 4.79$ min, 20 %; 13c, $T_R = 4.91$ min, 22 %; 13d, $T_R = 5.02$ min, 53.5 %).

Column chromatography on silica gel (1:1 followed by 1:2 hexane-ethyl acetate) yielded enriched mixtures of isomers **13a** and **13b** first, followed by a mixture of isomers **13b**, **13c** and **13d**, and, finally, a mixture (*c.a.* 2:5) of isomers **13c** and **13d**. Further preparative TLC(1:1 hexane-ethyl acetate) allowed isomers **13a** and **13b** to be isolated to some extent as colorless syrups. Attempts at isolating isomers **13c** and **13d** failed; spectral data were taken from a 2:5 mixture of isomers **13c** and **13d**.

13a: ¹H NMR (CDCl₃) δ : 4.11 (dd, 1H, J₅₅ = 8.9 and J₅₇₄ = 7.3 Hz, H₅), 3.89 (dd, 1H, J₅₅ = 8.9 and J₅₄ = 6.1 Hz, H₅), 3.76 (dd, 1H, J₄₅ = 7.3 and J₄₅ = 6.1 Hz, H₄), 3.41 (s, 1H, H₂), 3.12 and 2.92 (2s, 6H, NMe₂), 1.46, 1.39 and 1.30 (3s, 9H, CMe₂ and Me). ¹³C NMR (CDCl₃) δ : 166.0 (CO), 109.9 (OCO), 77.2 (C₄), 66.4 (C₅), 60.2 (C₃), 60.1 (C₂), 36.5 and 35.1 (NMe₂), 25.9 and 24.6 (C<u>Me₂</u>), 15.6 (Me). EIMS *m/z* (%): 214 (M⁺-15, 48), 157 (38), 128 (6), 112 (15), 102 (23), 99 (23), 85 (38), 72 (100), 43 (95).

13b: ¹H NMR (CDCl₃) δ : 4.05-3.95 (m, 2H, H_s and H_s), 3.74 (dd, 1H, J_{4,5} = 5.4 and J_{4,5} = 6.5 Hz, H₄), 3.41 (s, 1H, H₂), 3.07 and 2.97 (2s, 6H, NMe₂), 1.43, 1.42 and 1.25 (3s, 9H, CMe₂ and Me). ¹³C NMR (CDCl₃) δ : 166.1 (CO), 109.9 (OCO), 74.8 (C₄), 66.0 (C₅), 61.5 (C₃), 61.3 (C₂), 36.2 and 35.2 (NMe₂), 26.1 and 24.5 (C<u>Me₂</u>), 16.2 (Me). EIMS *m*/*z* (%): 214 (M^{*}-15, 46), 157 (55), 128 (8), 112 (6), 102 (20), 99 (29), 85 (20), 72 (100), 43 (82).

13c: ¹H NMR (CDCl₃) δ : 4.13-3.95 (m, 2H, H₅ and H₅), 3.85 (dd, 1H, J₄₅ = 4.7 and J₄₅ = 6.3 Hz, H₄), 3.68 (s, 1H, H₂), 3.07 and 2.96 (2s, 6H, NMe₂), 1.38, 1.33 and 1.23 (3s, 9H, CMe₂ and Me). ¹³C NMR (CDCl₃) δ : 166.2

(CO), 110.0 (OCO), 76.6 (C₄), 65.6 (C₅), 61.2 (C₃), 56.9 (C₂), 36.1 and 35.2 (NMe₂), 25.7 and 25.4 (C<u>Me₂</u>), 14.5
(Me). EIMS m/z (%): 214 (M*-15, 65), 172 (6), 128 (14), 112 (46), 102 (26), 85 (18), 72 (82), 43 (100).

13d: ¹H NMR (CDCl₃) δ : 4.10 (t, 1H, $J_{55} = J_{5,4} = 6.0$ Hz, H_5), 3.98 (t, 1H, $J_{5,5} = J_{5,4} = 6.0$ Hz, H_5), 3.90 (t, 1H, $J_{45} = J_{45} = 6.0$ Hz, H_4), 3.57 (s, 1H, H_2), 3.08 and 2.96 (2s, 6H, NMe₂), 1.42, 1.32 and 1.24 (3s, 9H, CMe₂ and Me). ¹³C NMR (CDCl₃) δ : 166.2 (CO), 110.0 (OCO), 77.6 (C₄), 65.9 (C₅), 61.3 (C₃), 58.3 (C₂), 36.1 and 35.2 (NMe₂), 26.1 and 24.7 (C<u>Me₂</u>), 13.1 (Me). EIMS *m/z* (%): 214 (M⁺-15, 66), 172 (6), 128 (14), 112 (38), 102 (19), 85 (18), 72 (92), 43 (100).

Anal. calcd. for C₁₁H₁₉NO₄: C, 57.62; H, 8.35; N, 6.11. Found: C, 57.43; H, 8.49; N, 6.14.

N,N-Dimethyl-3,6-anhydro-4,5-O-isopropylidene-D-glycero-D-manno-heptonamide (17)

A solution of 2,3-O-isopropylidene-D-ribofuranose¹⁰ (14, 1g, 5.26 mmol) in CHCl₃ (50 ml) was stirred at room temperature while ylide (2, 1.63 g, 11 mmol) was added sequentially until TLC analysis revealed depletion of the starting sugar. When the reaction was complete (*c.a.* 24 h), the solvent was evaporated and the residue purified by flash chromatography on silica gel using ethyl acetate to obtain 17 (1.23 g, 85%) as a colorless syrup. $[\alpha] = +$ 33.4° (c 0.43, CHCl₃). ¹H NMR (CDCl₃) &: 4.87 (dd, 1H, J₄₅= 6.2 and J₄₃= 3.8 Hz, H₄), 4.78 (t, 1H, J₂₃= 9.0 and J_{20H}= 9.0 Hz, H₂), 4.65 (dd, 1H, J₅₆= 1.2 and J₅₄= 6.2 Hz, H₅), 4.10 (dt, 1H, J₆₅= 1.2 and J₆₇= 6.0 Hz, H₆), 3.90 (dd, 1H, J₃₄= 3.8 and J₃₂= 9.0 Hz, H₂), 3.65 (d, 1H, J_{0H2}= 9.0 Hz, OH), 3.53 (d, 2H, J₇₆= 6.0 Hz, H₇ and H₇), 3.05 and 2.98 (2s, 6H, NMe₂), 1.49 and 1.33 (2s, 6H, CMe₂). ¹³C NMR (CDCl₃) &: 172.9 (CO), 112.1 (OCO), 85.0 (C₃), 83.1 (C₆), 82.0 (C₅), 81.2 (C₄), 65.2 (C₂), 61.8 (C₇), 36.8 and 35.7 (NMe₂), 26.5 and 24.5 (CMe₂). EIMS *m/z* (%): 260 (M^{*}-15, 6), 258 (M-17, 4), 257 (3), 213 (14), 203 (10), 168 (12), 152 (17), 124 (21), 114 (27), 103 (42), 102 (27), 72 (100), 59 (44).

Anal. calcd. for C1, H2, NO5: C, 52.35; H, 7.69; N, 5.08. Found: C, 52.05; H, 7.96; N, 4.93

N,N-Dimethyl-2,7-di-O-acetyl-3,6-anhydro-4,5-O-isopropylidene-D-glycero-D-manno- and D-glycero-D-allo-heptonamides (19 and 20)

A portion of the above crude reaction mixture (100 mg) in anhydrous pyridine (3 ml) was treated with Ac_2O (1 ml) at room temperature overnight. After the usual treatment, the crude product was purified by flash chromatography on silica gel with ethyl acetate, to give 20 (10 mg, 7.7%) and the more polar derivative 19 (112 mg, 85.8%) as colorless syrups.

19: $[\alpha] = +23.75^{\circ}(c\ 0.4, CHCl_3)$. ¹H NMR (CDCl_3) & 5.50 (d, 1H, J₂₃=9.5 Hz, H₂), 4.80 (dd, 1H, J₄₃=3.7 and J₄₅=6.0 Hz, H₄), 4.71 (d, 1H, J₅₄=6.0 Hz, H₅), 4.31 (dd, 1H, J₃₂=9.5 and J₃₄=3.7 Hz, H₃), 4.26 (dd, 1H, J₆₇=5.0 and J₆₇=4.5 Hz, H₆), 4.13 (dd, 1H, J₇₇=11.8 and J₇₆=4.5 Hz, H₇), 3.98 (dd, 1H, J₇₇=11.8 and J₇₆=5.0 Hz, H₇), 3.15 and 2.98 (2s, 6H, NMe₂), 2.10 and 2.06 (2s, 6H, 2 x CH₃CO), 1.47 and 1.33 (2s, 6H, CMe₂). ¹³C NMR (CDCl₃) & 170.4, 169.9 and 168.7 (3 x CO), 112.8 (OCO), 82.7 (C₅), 82.1 (C₆), 81.1 (C₃), 80.9 (C₄), 66.9 (C₂), 64.2 (C₇), 37.2 and 36.1 (NMe₂), 26.4 and 24.9 (C<u>Me₂)</u>, 20.9 and 20.6 (2 x <u>CH₃CO</u>). EIMS *m/z* (%): 359 (M⁺, 0.3), 344 (M⁺-15, 2), 300 (M⁺-59, 8), 197 (16), 181 (25), 137 (13), 81 (15), 72 (100), 43 (73).

20: $[\alpha] = +11.2$ ° (c 0.5, CHCl₃). ¹H NMR (CDCl₃) δ : 5.35 (d, 1H, J₂₃ = 4.2 Hz, H₂), 4.91 (dd, 1H, J₄₃ = 2.8 and J₄₅ = 6.3 Hz, H₄), 4.49 (dd, 1H, J₅₅ = 3.5 and J₅₄ = 6.3 Hz, H₃), 4.30 (dd, 1H, J₃₂ = 4.2 and J₃₄ = 2.8 Hz, H₄), 4.25-

4.04 (m, 3H, H_{6} , H_{7} and H_{7}), 3.08 and 2.95 (2s, 6H, NMe₂), 2.13 and 2.07 (2s, 6H, 2 x CH₃CO), 1.50 and 1.3 6H, CMe₂). ¹³C NMR (CDCl₃) δ : 170.6, 170.2 and 166.7 (3 x CO), 114.1 (OCO), 83.0 (C₆), 82.7 (C₃), 81.8 81.3 (C₄), 70.6 (C₂), 64.4 (C₇), 36.8 and 36.0 (NMe₂), 27.4 and 25.5 (CMe₂), 20.8 and 20.7 (2 x CH₃CO). I *m/z* (%): 344 (M⁺-15, 2), 300 (M⁺-59, 1), 197 (4), 181 (8), 137 (4), 81 (4), 72 (57), 43 (100).

Anal. calcd. for C₁₆H₂₅NO₈: C, 53.47; H, 7.01; N, 3.89. Found: C, 53.67; H, 7.28; N, 3.87.

In a separate experiment the pure isomer 17 (30 mg) was acetylated by the same procedure mentioned ϵ to give 19 (38 mg, 97%).

N,N-Dimethyl-2,3-anhydro-4,5:7,8-di-O-isopropylidene-D-*erithro*-L-*gluco*-octonamide (22), Dimethyl-3,6-anhydro-4,5:7,8-di-O-isopropylidene-D-*erithro*-L-*allo*-octonamide (24) and N,N-Dime 3,6-anhydro-4,5:7,8-di-O-isopropylidene-D-*erithro*-L-*manno*-octonamide (25).

A solution of 2,3:5,6-di-O-isopropylidene-D-mannofuranose¹¹ (21, 1g, 3.84 mmol) in CHCl₃ (90 ml stirred at room temperature while ylide (2, 1.5 g, 10.2 mmol) was added sequentially until TLC analysis rev depletion of the starting sugar. When the reaction was complete (*c.a.* 40 h), the solvent was evaporated amorphous white solid, which was shown by ¹H NMR to consist of three new compounds. Preliminary chromatography (1:2 hexane-ethyl acetate) provided an enriched mixture of each compound. The isomers isolated by further flash chromatography, followed by crystallization.

22: White solid (530 mg, 40%); m.p. 126-128°C (benzene); $[\alpha] = + 13.95^{\circ}$ (c 0.29, CHCl₃). ¹HNMR (Cl δ : 4.47 (dd, 1H, $J_{5,6} = 1.4$ and $J_{5,4} = 7.0$ Hz, H_5), 4.17 (dd, 1H, $J_{8,8} = 9.0$ and $J_{8,7} = 6.5$ Hz, H_8), 4.15 (ddd, 1H, 6.5, $J_{7,8} = 5.5$ and $J_{7,6} = 6.4$ Hz, H_7), 4.04 (dd, 1H, $J_{8,8} = 9.0$ and $J_{8,7} = 5.5$ Hz, H_8), 4.00 (dd, 1H, $J_{4,5} = 7.0$ and $J_{3,4}$ Hz, H_4), 3.70 (dt, 1H, $J_{6,7} = 6.4$, $J_{6,5} = 1.4$ and $J_{6,0H} = 6.4$ Hz, H_6), 3.55 (d, 1H, $J_{2,3} = 2.0$ Hz H_2), 3.53 (dd, 1H, $J_{3,2}$ and $J_{3,4} = 4.0$ Hz, H_3), 3.16 and 2.95 (2s, 6H, NMe₂), 1.52, 1.38, 1.37 and 1.34 (4s, 12H, 2 x CMe₂). ¹³CNMR (Cl δ : 167.1 (CO), 109.52 and 109.46 (2 x OCO), 77.3 (C₄), 76.8 (C₅), 75.9 (C₇), 70.3 (C₆), 67.8 (C₈), 57.6 (C₃), (C₂), 36.6 and 35.5 (NMe₂), 26.9 and 26.8 (C<u>Me₂</u>), 25.34 and 25.29 (C<u>Me₂</u>). EIMS *m/z* (%): 330 (M*-15, 5) (15), 244 (10), 186 (10), 185 (7), 128 (18), 103 (35), 102 (25), 101 (40), 85 (20), 72 (100). *Anal.* calcd. for C₁₆H₂₇NO₇; C, 55.64; H, 7.88; N, 4.05. Found: C, 55.69; H, 8.13; N, 4.00.

24: white crystals (504 mg, 38%); m.p. 165-167°C (benzene); $[\alpha]=-16.86^{\circ}$ (c 0.5, CHCl₃). ¹HNMR (Cl δ : 4.84-4.75 (m, 2H, H₄ and H₃), 4.53 (dd, 1H, J₂₃= 2.8 and J_{2.0H}= 6.3 Hz, H₂), 4.40-4.26 (m, 2H, H₆ and H₇), 4.04 (m, 3H, H₃, H₈ and H₈), 3.02 (s, 6H, NMe₂), 1.46, 1.43, 1.36 and 1.30 (4s, 12H, 2 x CMe₂). ¹³C NMR (Cl δ : 171.1 (CO), 112.3 and 109.0 (2 x OCO), 85.1 (C₃), 83.0 (C₆), 81.5 and 81.4 (C₄ and C₅), 73.8 (C₇), 70.7 66.6 (C₈), 36.6 and 36.2 (NMe₂), 26.7, 25.9, 25.1 and 24.3 (2 x CMe₂). EIMS *m/z* (%): 345 (M⁺, 0.3), 334 272 (30), 103 (75), 102 (42), 101 (52), 85 (25), 72 (100).

Anal. calcd. for C₁₆H₇₇NO₇; C, 55.64; H, 7.88; N, 4.05. Found: C, 55.31; H, 8.11; N, 3.86.

25: white solid (159 mg, 12%); m.p. 121-123°C. $[\alpha] = -18.18^{\circ}$ (c 0.1, CHCl₃). ¹H NMR (CDCl₃) & 4.88 1H, J₄₅ = 6.1 and J₄₃ = 3.5 Hz, H₄), 4.80 (d, 1H, J₂₃ = 9.1 Hz, H₂), 4.71 (dd, 1H, J₅₆ = 3.7 and J₅₄ = 6.1 Hz, H₅), (ddd, 1H, J₇₆ = 6.1, J₇₈ = 5.3 and J₇₈ = 6.0 Hz, H₇), 4.01 (dd, 1H, J₈₇ = 6.0 and J₈₈ = 8.5 Hz, H₈), 3.95 (dd, 1H, H₇₆), (ddd, 1H, J₇₆ = 6.1, J₇₈ = 5.3 and J₇₈ = 6.0 Hz, H₇), 4.01 (dd, 1H, J₈₇₇ = 6.0 and J₈₈ = 8.5 Hz, H₈), 3.95 (dd, 1H, H₇₆), (dd, 1H, J₇₆ = 6.1 Hz, H₈), 3.95 (dd, 1H, H₈₆), 3.95 (dd, 1H, H₈₆)

5.3 and $J_{g,g}$ = 8.5 Hz, H_g), 3.51 (dd, 1H, $J_{6,5}$ = 3.7 and $J_{6,7}$ = 6.1 Hz, H_g), 3.42 (dd, 1H, $J_{3,2}$ = 9.1 and $J_{3,4}$ = 3.5 Hz, H_3), 3.00 and 2.99 (2s, 6H, NMe₂), 1.39, 1.35 and 1.34 (3s, 3H, 3H and 6H, 2 x CMe₂). ¹³C NMR (CDCl₃) δ : 172.7 (CO), 112.4 and 108.9 (2 x OCO), 83.7 (C₃), 82.2 (C₆), 80.8 (C₄), 79.8 (C₅), 73.1 (C₇), 66.3 (C₉), 64.6 (C₂), 36.8 and 35.9 (NMe₂), 26.7, 25.7, 25.1 and 24.1 (2 x CMe₂). EIMS *m/z* (%): 345 (M⁺, 1), 330 (13), 272 (14), 186 (9), 185 (8), 115 (14), 103 (42), 102 (24), 101 (32), 85 (22), 72 (100).

Anal. calcd. for C₁₆H₂₇NO₇; C, 55.64; H, 7.88; N, 4.05. Found: C, 55.30; H, 8.18; N, 3.93

The following acetylated derivatives were prepared with a 92-95 % yield from 22, 24 and 25 respectively, using the procedure above described, all were purified by preparative TLC.

N,N-Dimethyl-6-O-acetyl-2,3-anhydro-4,5:7,8-di-O-isopropylidene-D-*erithro*-L-*gluco*-octonamide (26)

Colorless syrup; $[\alpha] = +29.69^{\circ}$ (c 0.32, CHCl₃). ¹HNMR (CDCl₃) δ : 5.25 (dd, 1H, $J_{6,5} = 2.4$ and $J_{6,7} = 6.6$ Hz, $H_{6,7}$), 4.45 (dd, 1H, $J_{5,4} = 6.8$ and $J_{5,6} = 2.4$ Hz, H_{5}), 4.21 (ddd, 1H, $J_{7,6} = 6.6$, $J_{7,8} = 6.7$ and $J_{7,8} = 6.0$ Hz, H_{7}), 4.08 (dd, 1H, $J_{4,3} = 4.6$ and $J_{4,5} = 6.8$ Hz, H_{4}), 3.98 (dd, 1H, $J_{8,7} = 6.0$ and $J_{8,8} = 8.8$ Hz, H_{8}), 3.87 (dd, 1H, $J_{8,7} = 6.7$ and $J_{8,8} = 8.8$ Hz, H_{8}), 3.53 (d, 1H, $J_{2,3} = 2.1$ Hz, H_{2}), 3.21 (dd, 1H, $J_{3,2} = 2.1$ and $J_{3,4} = 4.6$ Hz, H_{3}), 3.11 and 2.94 (2s, 6H, NMe₂), 2.08 (CH₃CO), 1.48, 1.35 and 1.32 (3s, 3H, 3H and 6H, 2 x CMe₂). ¹³C NMR (CDCl₃) δ : 170.4 and 166.4 (CH₃CO) and <u>CO</u>NMe₂), 109.7 and 109.5 (2 x OCO), 76.1 (C₄), 75.8 and 75.6 (C₅ and C₇), 69.6 (C₆), 66.4 (C₈), 55.6 (C₃), 49.3 (C₂), 36.5 and 35.9 (NMe₂), 26.4 and 25.5 (2 x CMe₂), 21.1 (CH₃CO). EIMS *m/z* (%): 372 (M⁺-15, 9), 314 (3), 114 (20), 101 (30), 72 (100).

N,N-Dimethyl-2-O-acetyl-3,6-anhydro-4,5:7,8-di-O-isopropylidene-D-erithro-L-allo-octonamide (27)

Colorless syrup; $[\alpha] = -6.9^{\circ}$ (c 0.36, CHCl₃); ¹H NMR (CDCl₃) δ : 5.29 (d, 1H, J_{2,3} = 3.6 Hz, H₂), 5.06 (d, 1H, J_{4,5} = 6.2, H₄), 4.81 (dd, 1H, J_{5,4} = 6.2 and J_{5,6} = 4.1 Hz, H₅), 4.32 (q, 1H, J_{7,8} = J_{7,8} = J_{7,6} = 6.0 Hz, H₇), 4.24 (d, 1H, J_{3,2} = 3.6, H₃), 4.10-3.99 (m, 3H, H₆, H₈), 3.08 and 2.93 (2s, 6H, NMe₂), 2.09 (s, 3H, CH₃CO), 1.44, 1.40, 1.32 and 1.30 (4s, 12H, 2 x CMe₂). ¹³C NMR (CDCl₃) δ : 169.9 and 166.6 (CONMe₂ and CH₃CO), 112.7 and 108.9 (2 OCO), 83.2 (C₃), 82.7 (C₆), 81.6 and 81.1 (C₄ and C₅), 73.7 (C₇), 72.3 (C₂), 66.2 (C₈), 36.9 and 36.0 (NMe₂), 26.8, 25.9, 25.0 and 24.4 (2 x CMe₂), 20.8 (CH₃CO). EIMS *m/z* (%): 372 (M⁺-15, 9), 314 (45), 168 (45), 145 (91), 103 (46), 102 (14), 101 (49), 85 (12), 72 (100), 43 (75).

N,N-Dimethyl-2-O-acetyl-3,6-anhydro-4,5:7,8-di-O-isopropylidene-D-*erithro*-L-*manno*-octonamide (28)

White solid; m.p. 134-137°C; $[\alpha] = -27.7^{\circ}$ (c 0.31, CHCl₃). ¹H NMR (CDCl₃) δ : 5.48 (d, 1H, J_{2,3}=9.5 Hz, H₂), 4.75-4.73 (m, 2H, H₄ and H₃), 4.32 (ddd, 1H, J_{7,8}= 6.0, J_{7,8}= 5.4 and J_{7,8}= 6.3 Hz, H₇), 4.0 (dd, 1H, J_{8,7}= 6.0 and J_{8,8}= 8.5 Hz, H₈), 3.94 (dd, 1H, J_{8,7}= 5.4 and J_{8,8}= 5.4 Hz, H₈), 3.86 (dd, 1H, J_{3,2}= 9.5 y J_{3,4}= 1.2 Hz, H₃), 3.52 (dd, 1H, J_{6,7}= 6.3 and J₆₅ <1 Hz, H₆), 3.10 and 2.94 (2s, 6H, NMe₂), 2.10 (CH₃CO), 1.46, 1.36, 1.33 and 1.30 (4s, 12H, 2 x CMe₇). ¹³C NMR (CDCl₄) δ : 169.9 and 168.7 (CONMe₇ and CH₂CO), 112.6 and 109.0 (2 x OCO), 82.0

 (C_{s}) , 80.5, 80.2 and 80.1 $(C_{3}, C_{4} \text{ and } C_{5})$, 73.0 (C_{7}) , 66.3 and 66.2 $(C_{2} \text{ and } C_{3})$, 36.9 and 36.0 (NMe_{2}) , 26.6, 25.8, 25.2 and 24.2 (2 x C<u>Me}_{2}), 20.6 $(CH_{3}CO)$. EIMS m/z (%): 387 $(M^{+}, 2)$, 372 $(M^{+}-15, 20)$, 330 (8), 314 (4), 229 (7), 194 (8), 168 (11), 152 (24), 145 (21), 103 (11), 102 (6), 101 (26), 72 (100), 43 (60).</u>

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REFERENCES

- 1.- Valpuesta Fernández, M.; Durante Lanes, P.; López-Herrera, F.J. Tetrahedron, 1990, 46, 7911-7922.
- (a) Ohrui, H.; Jones, J.H.; Moffatt, J.G.; Maddox, M.L.; Christensen, A.T.; Byram, S.K. J. Am. Chem. Soc. 1975, 97, 4602-4613. (b) Ohrui, H.; Emoto, S. J. Org. Chem. 1977, 42, 1951-1957.
- 3.- Pino González, M.S.; Domínguez Aciego, R.M.; López-Herrera, F.J. Tetrahedron, 1988, 44, 3715-3726.
- Buchanan, J.G.; Edgar, A.R.; Rawson, D.I.; Shahidi, P.; Wightman, R.H. Carbohydr. Res., 1982, 100, 75-86.
- 5.- Jung, M.E.; Shaw, T.J., J. Am. Chem. Soc., 1980, 102, 6304-6311.
- 6.- Ball, D.H., J. Org. Chem., 1966, 31, 220-223.
- 7.- Wiggins, L.F., J. Chem. Soc., 1946, 13-14.
- 8.- Olin, S.M., Methods Carbohydr. Chem., 1962, 1, 148-151.
- 9.- Hagen, S.; Anthonsen, T.; Kilaas, L., Tetrahedron, 1979, 35, 2583-2589.
- 10.- Hughes, N.A.; Speakman, P.R.H., Carbohydr. Res., 1965, 5, 171-175.
- 11.- Schmidt, O.T., Methods Carbohydr. Chem., 1963, 2, 318-319.